Solubility Rules for Common Ionic Compounds in Water Some Soluble Compounds Exceptions Group IA cations (Li+, Na+, K+. Rb+, No Exceptions

Cs+), NH4+ NO₃. ClO₄, CH₃COO, ClO₃

SO42-

Cl., Br : 1

Some Insoluble Compounds

52-

CO32-OH.

PO43-

Hydroxides of Group 1A cations, NH₄+, Ca2+, Sr2+, Ba2+

Sr2+, Ba2+

No Exceptions

Halides of Ag+, Hg22+, Pb2+

Sulfates of Sr2+. Ba2+, Hg22+, Pb2+

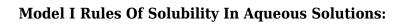
Sulfides of Group 1A cations, NH4+, Ca2+,

Carbonates of Group 1A cations and NH₄+

Phosphates of Group 1A cations and NH₄⁺

Model I Rules Of Solubility In Aqueous Solutions

Judd E. Hollander



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