

Solubility Rules for Common Ionic Compounds in Water

Some Soluble Compounds

Group 1A cations (Li^+ , Na^+ , K^+ , Rb^+ , Cs^+), NH_4^+

NO_3^- , ClO_4^- , CH_3COO^- , ClO_3^-

Cl^- , Br^- , I^-

SO_4^{2-}

Exceptions

No Exceptions

No Exceptions

Halides of Ag^+ , Hg_2^{2+} , Pb^{2+}

Sulfates of Sr^{2+} , Ba^{2+} , Hg_2^{2+} , Pb^{2+}

Some Insoluble Compounds

S^{2-}

Sulfides of Group 1A cations, NH_4^+ , Ca^{2+} , Sr^{2+} , Ba^{2+}

CO_3^{2-}

Carbonates of Group 1A cations and NH_4^+

OH^-

Hydroxides of Group 1A cations, NH_4^+ , Ca^{2+} , Sr^{2+} , Ba^{2+}

PO_4^{3-}

Phosphates of Group 1A cations and NH_4^+

Model I Rules Of Solubility In Aqueous Solutions

Judd E. Hollander



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