Name	
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Mixed Practice: Names and Formulas

First determine whether each of the following compounds is *ionic* (M + NM), polyatomic (3 or more atoms), or covalent (2 NM's). Then write the name for each compound.

1) NO ₂		
2) NaBr		
3) SiO ₂		
4) P ₂ Br ₄		
5) FeSO ₄		
6) SF ₆		
7) Li ₂ S		
8) MgBr ₂		
9) N ₂ S		
10) Be(OH)₂		
11) SO ₃		
12) Cu₂S		
13) BF ₃		
14) I ₂		
15) Pb ₃ (PO ₃) ₂		

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In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "Modeling Chemistry Unit 4 Ws Answers," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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