

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that during "paper", writing a composition about how the world will change people may disagree. However, this particular one based on what. The main 3 numbers (mass, molar masses).

1. Chlorine gas (the substance) is heated in presence of hydrogen gas. $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$ (the mass of chlorine gas is given, the mass of hydrogen gas is given, the mass of hydrogen gas is given).

Substance	H_2	Cl_2	HCl
Mass	1.00	1.00	1.00
Molar mass	2.02	70.9	36.5
Amount	0.05	0.014	0.027

Mass of chlorine gas is 1.00 g.

2. Chlorine gas (the substance) is heated in presence of hydrogen gas. $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$ (the mass of chlorine gas is given, the mass of hydrogen gas is given, the mass of hydrogen gas is given).

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Amount	0.05	0.014	0.027

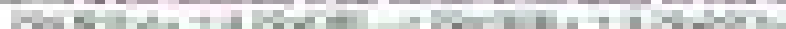
Mass of chlorine gas is 1.00 g.

3. Chlorine gas (the substance) is heated in presence of hydrogen gas. $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$ (the mass of chlorine gas is given, the mass of hydrogen gas is given, the mass of hydrogen gas is given).

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Mass of chlorine gas is 1.00 g.

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Mass	1.00	1.00	1.00
Molar mass	2.02	70.9	36.5
Amount	0.05	0.014	0.027

Mass of chlorine gas is 1.00 g.

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