

Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being in their relatively distinctive chemical behaviour, which associated particles must be from negatively charged ions (anions). However, when the elements react, molecules formed (molecules) are associated by their composition (they do not contain elements) as units. We will now examine the patterns that exist for the ratios in which these elements combine in order to describe the changes of the basic building blocks.

1. Write the formula and draw the particle diagram for each compound. The ratio of ions in each compound is given.

Atomic number Element	16 electrons 16 protons $^{16}\text{S}^{2-}$	12 electrons 12 protons $^{12}\text{Mg}^{2+}$	17 electrons 17 protons $^{17}\text{Cl}^{-}$	19 electrons 19 protons $^{19}\text{K}^{+}$
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particle



Diagram

Atomic number Element	16 electrons 16 protons $^{16}\text{S}^{2-}$	17 electrons 17 protons $^{17}\text{Cl}^{-}$	19 electrons 19 protons $^{19}\text{K}^{+}$
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particle



Diagram

Atomic number Element	16 electrons 16 protons $^{16}\text{S}^{2-}$	12 electrons 12 protons $^{12}\text{Mg}^{2+}$	17 electrons 17 protons $^{17}\text{Cl}^{-}$	19 electrons 19 protons $^{19}\text{K}^{+}$
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particle



Diagram

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A Loxley



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bei einer histaminintoleranz ist das verhältnis zwischen dem abbau von histamin und seiner produktion im körper beziehungsweise der zufuhr mit der nahrung nicht im gleichgewicht wird eine gewisse menge an histamin im körper überschritten kann es zu unterschiedlichen beschwerden kommen wie hoch diese menge ist ist individuell unterschiedlich

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bei einem teil der nicht allergisch bedingten nahrungsmittelunverträglichkeiten spielt der botenstoff histamineine bedeutende rolle entweder weil die auslösenden stoffe größere mengen histamin enthalten beziehungsweise freisetzen oder aber den raschen abbau von histamin im körper behindern zumindest in einem teil dieser fälle spricht

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