#### Nomenclature Worksheet 3 - Covalent (Molecular) Compounds

Part A: Name the following covalent	Part B: Write the Chemical Formula
compounds.	for each of the following compounds.
I. CO	carbon tetrafluorida
2. CO <sub>2</sub>	silicon dioxide
I. N <sub>2</sub> O <sub>1</sub>	dinitropen trisulfide
i. N <sub>3</sub> ,	4. phosphorus monoritride
s. NP	5. hydrogen gas
i. SC1 <sub>2</sub>	6. zurben disublide
2. P <sub>2</sub> O <sub>5</sub>	7. nitrogen trichkeride
i. Nile <sub>z</sub>	8. silicon tetrabromide
. Cl	9. curbon dioxide
0. CCI,	10. nitrogen triffsoride
I. PPs.,	1.1. berens trivalfide
2. PP <sub>3</sub>	12. sulphor trioxide
3. OS	13. setenium tetraffroride
4. SeF <sub>1</sub>	14. diphosphorus pentasulfide
5. Telle;	15. xenos setrafluoride
6. P <sub>2</sub> S <sub>5</sub>	16. sulibar didocernide
7. C <sub>3</sub> N <sub>a</sub>	17. carbon tetrachloride
S. F <sub>2</sub>	IK. onygen gur
9. CH <sub>4</sub>	19. fluorine gas
XX, PH:	20. dinitrogen tetroxide

# **Mixed Ionic Compound Homework**

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