



Chemistry

Chapter 10

Name _____

Stoich Worksheet 2

Stoichiometry Practice Problems 1

Do this on a separate sheet of paper. **Show all work and use units.** Round your answers to 3 significant digits.

1. What mass of iron is needed to react completely with 64.2 g of sulfur?



2. How much copper (I) oxide is needed to produce 75.0 g of copper (I) chloride?



3. How much hydrogen sulfite is produced when 178 g of sulfur dioxide reacts with water?



4. If 65.0 g of iron (II) sulfide is added to hydrogen chloride, what mass of hydrogen sulfide is produced?



5. Silver bromide is a precipitate that is formed when silver nitrate reacts with sodium bromide. What mass of the precipitate is produced if you start with 48.7 g of silver nitrate?



6. How much hydrogen nitrate is needed to react completely with 36.3 g of magnesium?



7. What mass of hydrogen can be produced when 82.1 g of zinc reacts with hydrogen sulfate?



For #8 & 9, you will need to use the density of oxygen gas is 1.43g/L.

8. What volume of oxygen is produced when 28.9g of water is decomposed?



9. What volume of oxygen is produced from the decomposition of 145.7g of potassium chlorate?



Mixed Stoichiometry Word Problems

J Rink



Mixed Stoichiometry Word Problems:

Reviewing **Mixed Stoichiometry Word Problems**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing.

Within the pages of "**Mixed Stoichiometry Word Problems**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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