

18.17 Application of Thermodynamic Principles

18.17.1

18.17

To power a hydroelectric system, two reservoirs having elevations that produce a constant pressure gradient, P , are joined under isothermal conditions. The system (see Fig. 18.17.1) consists of two large pipes, each having length L and cross-sectional area A . The pipes are connected at the bottom and the top. The system is initially at equilibrium. The system is then allowed to operate for a time t . The system is then allowed to operate for a time t . The system is then allowed to operate for a time t .

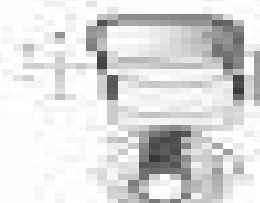


Figure 18.17.1: Hydroelectric system



Figure 18.17.2: Hydroelectric system

$$P = \frac{F}{A} = \frac{mg}{A} = \frac{\rho Vg}{A} = \frac{\rho Ahg}{A} = \rho gh$$

- What is the change in the enthalpy of the system? (The system is initially at equilibrium. The system is then allowed to operate for a time t . The system is then allowed to operate for a time t . The system is then allowed to operate for a time t .)
- What is the change in the entropy of the system? (The system is initially at equilibrium. The system is then allowed to operate for a time t . The system is then allowed to operate for a time t . The system is then allowed to operate for a time t .)
- What is the change in the internal energy of the system? (The system is initially at equilibrium. The system is then allowed to operate for a time t . The system is then allowed to operate for a time t . The system is then allowed to operate for a time t .)
- What is the change in the Gibbs free energy of the system? (The system is initially at equilibrium. The system is then allowed to operate for a time t . The system is then allowed to operate for a time t . The system is then allowed to operate for a time t .)
- What is the change in the Helmholtz free energy of the system? (The system is initially at equilibrium. The system is then allowed to operate for a time t . The system is then allowed to operate for a time t . The system is then allowed to operate for a time t .)
- What is the change in the enthalpy of the system? (The system is initially at equilibrium. The system is then allowed to operate for a time t . The system is then allowed to operate for a time t . The system is then allowed to operate for a time t .)
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