

quantities then express mass, length and time in terms of dimensions of these quantities.

Answer:

Principle of homogeneity is used for solving this problem.

$$[h] = [ML^2T^{-1}]$$

$$[c] = [LT^{-1}]$$

$$[G] = [M^{-1}L^3T^{-2}]$$

$$\text{Let } m = kc^ah^bG^c$$

Solving the above we get,

$$m = kc^{-3/2}h^{3/2}G^{-1/2} = k\sqrt{ch}/G$$

$$\text{Let } L = kc^ah^bG^c$$

Solving the above we get,

$$L = kc^{-3/2}h^{3/2}G^{1/2} = k\sqrt{hG}/c^2$$

$$\text{Let } T = kc^ah^bG^c$$

Solving the above we get,

$$L = kc^{-3/2}h^{3/2}G^{1/2} = k\sqrt{hG}/c^2$$

2.41. An artificial satellite is revolving around a planet of mass M and radius R , in a circular orbit of radius r . From Kepler's third law about the period of a satellite around a common central body, square of the period of revolution T is proportional to the cube of the radius of the orbit r . Show using dimensional analysis, that $T = k/R \sqrt{r/g}$ where k is a dimensionless constant and g is acceleration due to gravity.

Answer:

From Kepler's third law, we know that

$T^2 \propto r^3$ where T^2 is the square of time period of the satellite revolving around a planet and is proportional to the cube of the radius of the orbit r^3 .

$$T^2 \propto r^3$$

$$T \propto r^{3/2}$$

T depends on R and g

$$T \propto r^{3/2}g^aR^b$$

2.42. In an experiment to estimate the size of a molecule of oleic acid, 1mL of oleic acid is dissolved in 19mL of alcohol. Then 1mL of this solution is diluted to 20mL by adding alcohol. Now, 1 drop of this diluted solution is placed on water in a shallow trough. The solution spreads over the surface of water forming one molecule thick layer. Now, lycopodium powder is sprinkled evenly over the film we can calculate the thickness of the film which will give us the size of oleic acid molecule.

Read the passage carefully and answer the following questions:

a) Why do we dissolve oleic acid in alcohol?

b) What is the role of lycopodium powder?

c) What would be the volume of oleic acid in each mL of solution prepared?

d) How will you calculate the volume of n drops of this solution of oleic.

e) What will be the volume of oleic acid in one drop of this solution?

Answer:

a) Oleic is dissolved in the alcohol because it does not get dissolved in water.

b) With the help of lycopodium powder one can measure the area over which the oleic acid spreads as lycopodium powder clears the circular area when oleic acid is added.

c) 20mL of oleic acid contains 1mL of oleic acid.

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