11 to	Which compound he composition by mass	as the greatest percent	1.8	Which unit is used to thermal energy?
The state of	(1) BaS (2) CaS	(3) MgS (4) SrS		(1) gram (2) mole
UP!	Two molecules of HB	is released is absorbed	19 Judy PP	Under which condition pressure does a real gaideal gas? (1) low temperature and (2) low temperature and (4) high temperature and (4)
Second .	electron configuratio	The state of the s	20 Way	According to the kinetic ideal gas, all gas particles
本州	(1) earbon (2) magnesium	(4) oxygen	#18	(2) are in random, consta (2) are separated by very to their sizes
	bond?	scribes a <u>multip</u> le covalent		(3) have strong intermole (4) have collisions that de of the system
May	(1) Two electrons as (2) Four electrons as (3) Two electrons as (4) Four electrons as	re shared. e transferred.	21	Which mathematical exp heat of reaction for a che
Most	The electronegativi atoms in a molecul determine	by difference between the	way to	(1) (the heat of fusion) – (2) (the heat of vaporization) (the potential energy potential energy of the potential energy potential energy of the poten
451	(d) the polarity of atoms	on energy of the atoms the bond between the two an not be broken down by	22 122 4 17	At 101.3 kPa and 298 K which compound absorb of heat as the entire sam
Many	chemical means? (1) CO and He (2) CO and NH ₃	③ Xe and He Table (4) Xe and NHa S		(2) NaCl + 3.38 For a reaction at equilib increase the rates of the reactions?
17	(2) chemically comb	a mixture differ in density ese properties can be used to stances sine the substances reezing point of the mixture etrical conductivity of the		(1) a decrease in the reactants (2) a decrease in the surf an increase in the term of the forward reaction

- express an amount of Table
 - (4) pascal
- (3) joule
- is of temperature and is behave most like an
- low pressure
 - high pressure
 - d low pressure
 - d high pressure
 - molecular theory for an
- mt, straight-line motion small distances relative
 - scular forces
 - serease the total energy
 - mession represents the unical reaction?
 - the heat of vaporization)
 - on) (the heat of fusion) of the products) - (the
 - e reactants)
 - of the reactants) (the re products)
- a 1.0-mole sample of s the greatest amount ple dissolves in water?
- - rium, which change cun e forward and reverse
 - concentration of the
 - see area of the products
 - aperature of the system
 - ctivation energy of the

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