

Chemical bonding - answers

1. Atoms form molecules by sharing electrons and those are known as covalently bonded molecules. As same, by donating electrons to non-metals from metals the ionic compounds are forms.

a) Group the following compounds – ionic or covalent.

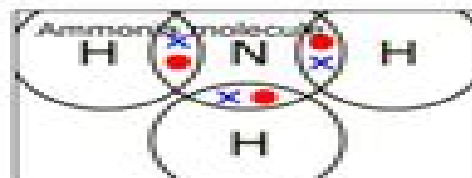
Carbon dioxide Magnesium Aluminium oxide water sulfur trioxide
Potassium iodide methane ethanol oxygen ammonium chloride ammonia gas
Diamond copper nitrate silicon dioxide sodium chloride hydrogen chloride gas

Covalent compounds	Ionic compounds
Carbon dioxide	Aluminium oxide
Water	Potassium iodide
Sulfur trioxide	Copper nitrate
Methane	Sodium chloride
Ethanol	
Oxygen	
Ammonia gas	
Silicon dioxide	
Hydrogen gas	

b). Simple molecular compounds can be formed by sharing electrons only. Give 3 examples for simple molecular compounds.

1.....methanol 2.....sulfur dioxide 3.....hydrogen sulfide gas (any covalent compound)

c. Draw dot cross diagrams to show the bonding in: i) hydrogen chloride molecule ii) ammonia molecule



d) Explain why the bond in a nitrogen molecule is called a triple bond.

Two nitrogen atoms share three electrons from each atom to form three covalent bonds.

e). How many electrons are need to form a covalent bond?

Two electrons / a pair of electrons

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AN Whitehead



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Table of Contents Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers

1. Understanding the eBook Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - The Rise of Digital Reading Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - Personalized Recommendations
 - Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers User Reviews and Ratings
 - Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers and Bestseller Lists
5. Accessing Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers Free and Paid eBooks
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 - Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers eBook Subscription Services
 - Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers Budget-Friendly Options
6. Navigating Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers Compatibility with Devices
 - Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - Highlighting and Note-Taking Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - Interactive Elements Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 8. Staying Engaged with Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 9. Balancing eBooks and Physical Books Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - Setting Reading Goals Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - Fact-Checking eBook Content of Holt Mcdougal Activity L1 Chemical Bonding Vocab Activity Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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