

Haynes

Restoration Manual



Morris Minor and 1000



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 cesium cs has the following abundance cs 132 20 0 cs 133 75 3 cs 134 4 7 weighted average atomic mass 132×0.200 133×0.752 134×0.047 132 85 amu

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 how you would determine the average atomic mass of an element section 2 data table don't forget calibration and units total
 all beans 100 beanium isotope blackium beanium isotope brownium beanium

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 atoms beans 2 minus 3 average mass of 1 bean 4 divided by 1 of beans 1 a b c $\times 100$ section 3 calculate the average atomic
 mass of beanium

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