

## **Stoichiometry and The Mole – Review Guide**

**Remember:**

- 1 mole =  $6.02 \times 10^{23}$  atoms, molecules, formula units
- 1 mole = grams (of molar mass)
- 1 mole = 22.4 liters (L) of gas



- 1) The process used to predict the amount of a product that will be formed is stoichiometry
- 2) If you know the mass of a substance, you can find the number of moles by dividing it by the...  
molar mass of substance
- 3) If you know the number of particles (atoms) of a substance, you can find the number of moles by dividing it by...  
Avogadro's Number
- 4) Summarize the steps of a gram-to-gram stoichiometry problem:  
  
Grams A → Moles A → Moles B → Grams B
- 5) The essential stoichiometry problem needed to convert from one chemical to another is:  
Mole to Mole
- 6) Two reactants undergo a chemical reaction. At the end of the reaction, Chemical A still exists in the container, while Chemical B has been completely used up. Which is the limiting reactant?  
Chemical B
- 7) What is a mole? A unit of measurement used to count atoms/molecules.
- 8) What is Avogadro's Number?  $6.02 \times 10^{23}$
- 9) What does Avogadro's number tell us? The number of atoms or molecules in a mole.

# Learning Guide To Molar Relationship Answers

**Emma Kirk**

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular gradient shape partially visible behind it.

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**Readers' Guide to Periodical Literature** Anna Lorraine Guthrie,Bertha Tannehill,Neltje Marie Tannehill Shimer,1910

**Chemists' Guide to Effective Teaching** Norbert J. Pienta,Melanie M. Cooper,Thomas J. Greenbowe,2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry      *Teaching and Learning STEM* Richard M. Felder,Rebecca Brent,2016-02-22 Rethink traditional teaching methods to improve student learning and retention in STEM Educational research has repeatedly shown that compared to traditional teacher centered instruction certain learner centered methods lead to improved learning outcomes greater development of critical high level skills and increased retention in science technology engineering and mathematics STEM disciplines Teaching and Learning STEM presents a trove of practical research based strategies for designing and teaching STEM courses at the university community college and high school levels The book draws on the authors extensive backgrounds and decades of experience in STEM education and faculty development Its engaging and well illustrated descriptions will equip you to implement the strategies in your courses and to deal effectively with problems including student resistance that might occur in the implementation The book will help you Plan and conduct class sessions in which students are actively engaged no matter how large the class is Make good use of technology in face to face online and hybrid courses and flipped classrooms Assess how well students are acquiring the knowledge skills and conceptual understanding the course is designed to teach Help students develop expert

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Exceptional Child Education Resources ,1988

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