

Chapter 17: Thermochemistry*Section 1: The Flow of Energy – Heat and Work*

1. What is thermochemistry? **The study of the energy changes that occur in changes of state or during chemical reactions.**
2. Heat always flows from a **warmer** object to a **cooler** object.
3. Given that $1 \text{ cal} = 4.184 \text{ J}$:
 - a. How many calories are in 325.9 J ? **77.89 cal**
 - b. How many J are in 34.7 cal ? **145.18 J**
4. What is the Law of Conservation of Energy (aka the First Law of Thermodynamics)? **States that in chemical and physical processes, energy is neither created nor destroyed – it is transferred or conserved.**
5. In an exothermic reaction:
 - a. Heat is **released** / absorbed
 - b. q is **positive** / **negative**
 - c. ΔH is **positive** / **negative**
 - d. Heat is a **reactant** / **product**
6. In an endothermic reaction:
 - a. Heat is **released** / **absorbed**
 - b. q is **positive** / **negative**
 - c. ΔH is **positive** / **negative**
 - d. Heat is a **reactant** / **product**
7. Diagram **A** represents an exothermic reaction, whereas diagram **B** represents an endothermic reaction:
8. What is the difference between heat capacity and specific heat capacity? **Specific heat is the amount of energy to increase the temperature of 1g of substance by 1°C**
9. The temperature of a 34.1 g sample of a substance increases from 29°C to 51°C when the sample absorbs 912 J of heat. What is the specific heat of the substance? **$1.126 \text{ J/g}^\circ\text{C}$**
10. When 867 J of heat are added to 9.3 g of a substance at STP, the temperature increases from 0°C to 14°C . What is the specific heat of the substance? **$6.66 \text{ J/g}^\circ\text{C}$**

Holt Modern Chemistry Chapter 17 Review Answers

Nicholas D. Tzimopoulos



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Books in Print Supplement**, 2002 **Holt Chemistry** Ralph Thomas Myers, 2004 Modern Chemistry Nicholas D. Tzimopoulos, 1993 **Chemistry and Industry Review**, 1928 Biochemistry Review Morris Silverman, 1982 **The Education Index**, 1938 Light of the Mind, Light of the World Spencer Klavan, 2024-10-15

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Humanity is not a mistake
For centuries a grim anti human outlook has taken hold of the public imagination teaching us all to view ourselves as random products of a cruel and uncaring natural world
Today from apocalyptic environmentalism to twisted eugenics and dystopian bionic augmentation movements are rising around the world to dispense with humanity or subordinate it to a pitiless mechanical logic
For many it has come to seem as if the human spirit is obsolete
religious faith is illusory and mankind is destined to be extinguished or surpassed
Some might even see the end of humanity as a good thing
But that is not our future
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