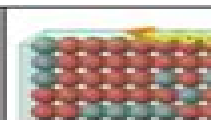


Activity B:
Measuring
half-life

Get the Gizmo ready:

- Click **Reset**.
- Select **Isotope A** from the left drop-down menu.
- Check that **Theoretical decay** is selected.



Introduction: Different **isotopes** of the same element have the same number of protons but different numbers of **neutrons** in the nucleus. Some isotopes are radioactive.

Question: How do we find the half-life of a radioactive isotope?

1. Observe: Select the **GRAPH** tab, and click **Play**. Based on the graph, what is your estimate of the half-life of isotope A?

15

2. Measure: Turn on the **Half-life probe**. Use the probe to measure how long it takes for exactly one-half of the original radioactive atoms to decay.

What is the exact half-life of isotope A?

31

3. Collect data: In the first row of the table below, write how many seconds represent one half-life, two half-lives, and so forth. On the next row, predict the number of radioactive atoms that will be present at each time. Then use the probe to find the actual values.

Half-life	0	1	2	3	4	5
Time (seconds)		31	62	94	125	155
Predicted # radioactive atoms	128	64	32	16	8	4
Actual # radioactive atoms	128	64	32	16	8	4

4. Calculate: Calculate the percentage of radioactive atoms that are left after each half-life.

Half-life	0	1	2	3	4	5
Percentage radioactive atoms	100	50	25	12.5	6.25	3.125

5. Apply: Suppose you found a material in which 12.5% of the original radioactive atoms were present. If the half-life is 47 years, how old is the material?

around 200 years of age

Half Life Lab Answer Key

G Thomas

A red circular graphic with a gradient, appearing as a stylized arrow or a light effect, positioned to the right of the 'G Thomas' text.

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Table of Contents Half Life Lab Answer Key

1. Understanding the eBook Half Life Lab Answer Key
 - The Rise of Digital Reading Half Life Lab Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Half Life Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Half Life Lab Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Half Life Lab Answer Key
 - Personalized Recommendations
 - Half Life Lab Answer Key User Reviews and Ratings
 - Half Life Lab Answer Key and Bestseller Lists
5. Accessing Half Life Lab Answer Key Free and Paid eBooks
 - Half Life Lab Answer Key Public Domain eBooks

- Half Life Lab Answer Key eBook Subscription Services
- Half Life Lab Answer Key Budget-Friendly Options
- 6. Navigating Half Life Lab Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Half Life Lab Answer Key Compatibility with Devices
 - Half Life Lab Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Half Life Lab Answer Key
 - Highlighting and Note-Taking Half Life Lab Answer Key
 - Interactive Elements Half Life Lab Answer Key
- 8. Staying Engaged with Half Life Lab Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Half Life Lab Answer Key
- 9. Balancing eBooks and Physical Books Half Life Lab Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Half Life Lab Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Half Life Lab Answer Key
 - Setting Reading Goals Half Life Lab Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Half Life Lab Answer Key
 - Fact-Checking eBook Content of Half Life Lab Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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