Density Worksheet Physical Science D=m/V

Substance	Dennity (g/cm+)	Substance	(g/cm²)
Chorgen.	0.00033	Alternation	9.79
Hydrogen	0.000084	Iron	7.87
Hithared	0.785	Copper	8.96
Beniese	0.880	Silver	140.5
Water	1.000	Lead	31-34
Mograpoisasa	1/74	Monutary	13-6
Salt (endium- chhoride)	2.16	Godd	19.38

 The ratio of an object's mass to its ______ is called the density of the object.

 A kilogram of lead occupies a much smaller volume than a kilogram of water, because ______ has a reach higher density.

 For the masses and volumes indicated, calculate the density in grams per cubic centimeters.

- a. mass = 453 g; volume = 225 cm3
- b. mass = 5.0 g; volume = 10.0 cm3.
- c. mass = 26.1 g; volume = 2.0 mL

4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.

A cube of metal weighs 1450 g and displaces 542 mL of water when immersed.
 Calculate the density of the metal.

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M Mark

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