

Radioactivity and Half-Lives Review Worksheet

1. What is the difference between radioactivity and nuclear radiation?

Radioactivity is the process of nuclear decay (an unstable isotope emitting energy/a particle to become more stable). Nuclear radiation is that energy or particle being emitted.

2. Which type of radiation can be blocked by---

- a. A piece of paper? **alpha**
- b. A piece of wood? **beta**
- c. A large block of concrete? **gamma**

Extension opportunity: Why? *What do we know about the size of each type of particle that affects their ability to pass through materials?* **Alpha particles are the largest, so they move the slowest and cannot penetrate objects as easily. Gamma particles are the smallest and have no charge, so they penetrate more easily.**

3. What is a half-life?

The time it takes for half of the material you started with to decay.

4. If we start with 400 atoms of a radioactive substance, how many would remain after one half-life? **200** After 2 half-lives? **100** After 3 half-lives? **50** After 4 half-lives? **25**

5. If we start with 16 grams of a substance, how much will remain after 3 half-lives?
 $16 \rightarrow 8 \rightarrow 4 \rightarrow 2$ **2 grams** after 3 half-lives.

6. A sample of carbon-14 has been decaying for 22,920 years and is now 35.0 grams. What was the size of the original sample? (The half-life of carbon-14 is 5,730 years)
 $22920/5730 = 4$ half lives *You will need to work backwards*
 $35 \rightarrow 70 \rightarrow 140 \rightarrow 280 \rightarrow$ **560. grams**

7. How old is a skeleton sample if the current amount of carbon in the bones is 3.125%? (Assume you started with 100%)
 $100\% \rightarrow 50 \rightarrow 25 \rightarrow 12.5 \rightarrow 6.25 \rightarrow 3.125$ 5 half lives $\times 5,730 =$ **28,650 years**

8. If you started with 120 grams of a radioactive substance, and now you have 15 grams, how many half-lives have passed?
 $120 \rightarrow 60 \rightarrow 30 \rightarrow 15$ **3 half-lives**

9. The half-life of isotope X is 2.0 years. How many years would it take for a 4.0 mg sample of isotope X to decay and have only 0.50 mg of it remain?
 $4.0 \rightarrow 2.0 \rightarrow 1.0 \rightarrow 0.50$ 3 half lives $\times 2$ years = **6.0 years**

Half Life Calculations Worksheet Answer Key

**Meta Brown Seltzer, Joyce L.
Mulholland, Joyce M. Mulholland**

Half Life Calculations Worksheet Answer Key:

Drug Calculations Meta Brown Seltzer, Joyce L. Mulholland, Joyce M. Mulholland, 2004 This resource provides all the information explanation and practice needed to competently and confidently calculate drug dosages The content features extensive coverage of the ratio and proportion method of drug calculation in a full color workbook including a logical step by step format for all problems and a proof step in the answer key to ensure that the users understand each calculation Each copy is packaged with the Daugherty Romans Dosages and Solutions CD ROM companion *Statistical Tools For Managers (using Ms Excel)* Apte, 2009 *Teacher File Year 8/1* David Baker, 2001 These resources provide invaluable support within the Key Maths series for all mathematics teachers whether specialists or non specialist experienced or new to the profession

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Half Life Calculations Worksheet Answer Key**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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