

Mole and Stoichiometry Mini-Lab

7. Write and balance the equation for the reaction between aluminum and water to produce aluminum oxide and hydrogen gas.



8. Using the moles of the aluminum can in #1 as the GIVEN, how many moles of hydrogen gas can be produced? Assume water is available in excess. UNKNOWN: mol H_2 GIVEN: mol Al (answer to #1)

9. Using the moles of water from #4 as the GIVEN with aluminum in excess, calculate the number of moles of aluminum oxide that can be produced.



Mini Lab 1 apply Stoichiometry

Rosina Ehmann



Mini Lab 1 apply Stoichiometry:

Mini Lab 1apply Stoichiometry Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Mini Lab 1apply Stoichiometry**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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