

Lewis Dot Structures of Atoms and Ions

Why?

The chemical properties of an element can be based on the number of electrons in the outer shell of the atoms. We can determine the structure of atoms and molecules by using the Lewis dot structures. Lewis dot structures are used to represent the valence electrons of an atom or molecule. The Lewis dot structures are used to represent the valence electrons of an atom or molecule. The Lewis dot structures are used to represent the valence electrons of an atom or molecule. The Lewis dot structures are used to represent the valence electrons of an atom or molecule.

Learning Objectives

1. Draw Lewis dot structures for elements in groups 1, 2, 13, 14, 15, 16, 17, and 18.

Success Criteria

1. Accurately draw Lewis dot structures for elements in groups 1, 2, 13, 14, 15, 16, 17, and 18.

Resources

1. Periodic table of elements showing valence electrons.
2. Lewis dot structures of elements in groups 1, 2, 13, 14, 15, 16, 17, and 18.

Steps/Concepts

1. The structure of the Lewis dot structures is as follows:

Procedure/Steps

1. Draw the Lewis dot structures of elements in groups 1, 2, 13, 14, 15, 16, 17, and 18.
2. Draw the Lewis dot structures of elements in groups 1, 2, 13, 14, 15, 16, 17, and 18.
3. Draw the Lewis dot structures of elements in groups 1, 2, 13, 14, 15, 16, 17, and 18.
4. Draw the Lewis dot structures of elements in groups 1, 2, 13, 14, 15, 16, 17, and 18.
5. Draw the Lewis dot structures of elements in groups 1, 2, 13, 14, 15, 16, 17, and 18.

Task

Draw the Lewis dot structures of

elements in groups 1, 2, 13, 14, 15, 16, 17, and 18.

Key Questions

1. What is the purpose of Lewis dot structures?

2. How many valence electrons are there in the Lewis dot structures?

Lewis Dot Structures Of Atoms And Ions Pogil Answers

Victor M. Corman



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