

ELECTROCHEMISTRY

A. Matching

Match each term in Column B to the correct description in Column A.

	Column A	Column B
1	a conductor in a circuit that carries electrons to or from a substance other than a metal	anode
2	one part of a voltaic cell, in which either oxidation or reduction occurs	electrode
3	a voltaic cell in which the electrolyte is a paste	cathode
4	the electrode at which reduction occurs	dry cell
5	a voltaic cell in which a fuel undergoes oxidation and from which electrical energy is obtained continuously	battery
6	an electrochemical cell that is used to convert chemical energy into electrical energy	electrolysis
7	a tube containing a conducting solution	voltaic cell
8	a group of cells that are connected together	half-cell
9	the process in which electrical energy is used to bring about a chemical change	salt bridge
10	the electrode at which oxidation occurs	fuel cell

B. Multiple Choice

Choose the best answer and write its letter on the line.

- 0**
11. In the electrolysis of brine, the substance produced at the cathode is
a. chlorine.
b. oxygen.
c. sodium.
d. hydrogen.
12. Which of the following describes a dry cell?
a. It can be recharged many times.
b. The graphite rod does not undergo reduction, even though it is the cathode.
c. It contains concentrated sulfuric acid.
d. all of the above.

High School Electrochemistry Study Guide Answers

Joacim Rocklöv



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