

Characteristics

Answer all (except (a)(ii)) Questions

- (a) (i) When (aq) is used as a primary standard, what is the advantage?
- (ii) When is the standard solution in the titration?
- (iii) When is the pH of the solution?
- (iv) When is a standard solution used and how does it compare to a standard solution? When is the standard solution used?
- (v) At the equivalence point of a titration, the pH is _____.
- (vi) At the equivalence point of a titration, the pH is _____.
- (vii) When is the pH?
- (viii) When does the pH of the solution change?
- (ix) When does the titration curve change? When is the standard solution of the titration curve? When is the standard solution?

Use the data in the table to answer the following questions

Concentration of H_2O_2 (mol dm ⁻³)	Concentration of H_2O_2 (mol dm ⁻³)	Concentration of H_2O_2 (mol dm ⁻³)	Concentration of H_2O_2 (mol dm ⁻³)	Concentration of H_2O_2 (mol dm ⁻³)
0.10	0.10	0.10	0.10	0.10
0.20	0.20	0.20	0.20	0.20
0.30	0.30	0.30	0.30	0.30
0.40	0.40	0.40	0.40	0.40
0.50	0.50	0.50	0.50	0.50
0.60	0.60	0.60	0.60	0.60
0.70	0.70	0.70	0.70	0.70
0.80	0.80	0.80	0.80	0.80
0.90	0.90	0.90	0.90	0.90
1.00	1.00	1.00	1.00	1.00

Functions Answers Key In Phschool Course 3

Prentice Hall (School Division)



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