



High School Auto Shop Class Workbook

**California State Polytechnic College
(San Luis Obispo)**

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web it is a small step from mole mass calculations to mass mass calculations if we start with a known mass of one substance in a chemical reaction instead of a known number of moles we can calculate the corresponding masses of other substances in the reaction

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web jul 4 2022 to calculate the mass of titanium metal that you can obtain multiply the number of moles of titanium by the molar mass of titanium 47.867 g/mol moles ti mass ti times molar mass ti $4.12 \text{ mol ti} \times 47.867 \text{ g/mol ti} = 197 \text{ g ti}$ notag

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