

acid) pH 1			hydrogen carbonate ('bicarb', baking soda) pH 8	13-14
HCl hydrochloric acid (in the lab is same as your stomach!) pH 1	CH ₃ COOH acetic/ethanoic acid (vinegar)	NaCl sodium chloride (salt water) pH 7	NH ₃ ammonia pH 11	KOH potassium hydroxide pH 13-14
HNO ₃ nitric acid pH 1	Fruit juices eg oranges and lemons contain citric acid pH 2-3	MgSO ₄ magnesium sulphate (Epsom salts) pH 6.5-7.0	Na ₂ CO ₃ sodium carbonate (washing soda) pH 11	oven cleaner (may contain NaOH) pH can be >12
	rain water (pH 5.5 normal, down to 3.5 when polluted)	C ₆ H ₁₂ O ₆ sugar pH 7	toothpaste pH 8	
	milk pH 6	C ₂ H ₅ OH ethanol (alcohol) pH 7	bleach pH 11	
	wine/beer pH 6		Ca(OH) ₂ calcium hydroxide (limewater) pH 10	
			Mg(OH) ₂ magnesium hydroxide ('milk of magnesia') pH 10	

Form Chemistry Notes

James W. McLaughlin



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