

Introduction to Engineering Design (IED)

End of Course Exam Study Guide

Unit 1: Design Process – 15% of Exam

1. How might one solve the most possible solutions to a problem?
2. What is the most effective way to generate potential solutions to a problem? How many alternate solutions should you generate?
3. What is an engineer? What types should the engineers do?
4. Why keep an Engineering Notebook?
5. What are the 4 basic functions for the Engineering Notebook?
6. Explain the process of problem-solving. How does the knowledge of problem-solving help in creating a solution?
7. Why would the ability to create potential solutions make a person a more independent designer?
8. What is the purpose of sketching your ideas?
9. Define and differentiate iteration and innovation.



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Introduction to Engineering Design Sven G. Bilén, 2001 United States Air Force Academy United States Air Force Academy, 1972 Curriculum Handbook with General Information Concerning ... for the United States Air Force Academy United States Air Force Academy, 2004 **Annual Catalogue** United States Air Force Academy, 1985 **Annual Catalog - United States Air Force Academy** United States Air Force Academy, 1971 *Transport Phenomena for Chemical Reactor Design* Laurence A. Belfiore, 2003-04-11 Laurence Belfiore's unique treatment meshes two mainstream subject areas in chemical engineering transport phenomena and chemical reactor design Expressly intended as an extension of Bird Stewart and Lightfoot's classic *Transport Phenomena* and Froment and Bischoff's *Chemical Reactor Analysis and Design* Second Edition Belfiore's unprecedented text explores the synthesis of these two disciplines in a manner the upper undergraduate or graduate reader can readily grasp *Transport Phenomena for Chemical Reactor Design* approaches the design of chemical reactors from microscopic heat and mass transfer principles It includes simultaneous consideration of kinetics and heat transfer both critical to the performance of real chemical reactors Complementary topics in transport phenomena and thermodynamics that provide support for chemical reactor analysis are covered including Fluid dynamics in the creeping and potential flow regimes around solid spheres and gas bubbles The corresponding mass transfer problems that employ velocity profiles derived in the book's fluid dynamics chapter to calculate interphase heat and mass transfer coefficients Heat capacities of ideal gases via statistical thermodynamics to calculate Prandtl numbers Thermodynamic stability criteria for homogeneous mixtures that reveal that binary molecular diffusion coefficients must be positive In addition to its comprehensive treatment the text also contains 484 problems and ninety six detailed solutions to assist in the exploration of the subject Graduate and advanced undergraduate chemical engineering students professors and researchers will appreciate the vision innovation and practical application of Laurence Belfiore's *Transport Phenomena for Chemical Reactor Design*

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