

**Test: Tuesday 2/18/20**

*This guide is not the only thing you should use to study. It does not provide you with everything you need. You should also rely on your **Chemical Interactions** textbook, homework, and classroom notes. Use everything you can for the best results.*

***\*\*Study your Solubility and Solutions Quiz***

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**Topic 1: Understanding Solutions – Textbook pg. 84-89; Binder pg 6-8**

1. *What happens to the particles of a solid when dissolved in water? What are the parts of that solution?*
  - *Particles of the solid leave each other and become surrounded by particles of water.*
  - *Ionic solids in water can conduct electricity; Molecular solids cannot!*
  - *Parts of the solution: Solvent:*
    - *Solvent = the largest amount; does the dissolving*
    - *Solute = the smaller amount; gets dissolved*
2. *Compare and contrast solutions, colloids, and suspensions.*

3. *Give an example of each (solutions, colloids, suspensions).*

- *Solutions = Air; saltwater*
- *Colloids = Fog; milk; Gelatin*
- *Solutions = Oil and Vinegar; Pepper/Water*

4. *How do solutes affect a solvent when added? Explain. (3 ways)*

- a. *Lower Freezing Point – Makes it harder for water molecules to form crystals of solid ice*
- b. *Increase Boiling Point – Makes it harder for water molecules to escape*
- c. *Ionic Solutes are able to conduct electricity in water – negative and positive ions are attracted to water molecules*

Contrast		
Solutions	Colloids	Suspensions
<i>*Same properties throughout *Molecules are too small to be seen **Can be made from solids, liquids, or gases</i>	<i>*Larger particles than a solution but still too small to be seen. **Able to scatter a beam of light</i>	<i>Particles can be seen and easily separated by settling or filtration DOES NOT have the same properties throughout</i>

**Topic 2: Concentration and Solubility – Textbook pg. 92-97; Binder pg 9-12**

5. *What is the difference between a saturated and an unsaturated solution? Give examples.*

- *Saturated - add so much solute that no more dissolves*
- *Unsaturated – Able to dissolve more solute*

6. *Explain the difference between a dilute and a concentrated solution. How is this different from a saturated and unsaturated solution?*

- *Dilute - Mixture that has only a little solute dissolved in a solvent*
- *Concentrated – Has a lot of solute dissolved in the solvent*
- *Saturated and Unsaturated solutions compares the ability of a solute to dissolve.*

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