

Intro. to Chemical Reactions

Name _____

Date _____

Section _____

Formula Review

1. Write the formulas for the following compounds:

- a. aluminum phosphate _____ b. sodium phosphate _____
 c. copper(II) phosphate _____ d. sodium sulfate _____

2. Write the formulas for the following compounds:

- a. CaCl_2 _____ b. FeCl_3 _____
 c. FeCl_2 _____ d. Cu_2S _____

Chemical Reactions: Virtual Energy Table

Write down observations for the reaction as it occurs:



There is no observable gas or color change when the two solutions are mixed.

Experiment 1 involves the reaction of hydrochloric acid and sodium hydroxide (also known as NaOH).

Write a balanced chemical equation for the reaction between sodium hydroxide and hydrochloric acid. Label the phases of the products.

Chemical Table Data Collection

3. Write the following reaction under the correct label (gas, precipitate):



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Padhraic Smyth



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