

7th Grade Science Vocabulary Quiz 19

1. B Coldest layer; meteors burn up here.
A.stratosphere B.mesosphere C.radiation D.thermosphere
2. C Gases such as carbon dioxide and methane, which absorb and give off infrared radiation as part of the Greenhouse Effect.
A.humidity B.mesosphere C.greenhouse gases D.ozone
3. A The amount of water vapor in the air.
A.humidity B.air pressure C.convection D.atom
4. B The force of air molecules pushing on an area.
A.mesosphere B.air pressure C.thermosphere D.air mass
5. B Basic unit of matter; smallest component of an element.
A.greenhouse gases B.atom C.nitrogen D.mesosphere
6. D Clear, dry layer; contains ozone layer.
A.oxygen B.atom C.greenhouse gases D.stratosphere
7. B The transfer of heat energy from one substance to another through direct contact.
A.stratosphere B.conduction C.greenhouse gases D.thermosphere
8. D A colorless, odorless, tasteless nonflammable gas; 21% of atmosphere.
A.mesosphere B.nitrogen C.thermosphere D.oxygen
9. A A colorless, odorless, tasteless gas; 78% of atmosphere.
A.nitrogen B.mesosphere C.air pressure D.radiation
10. D A gas molecule that consists of three oxygen atoms.
A.oxygen B.mesosphere C.nitrogen D.ozone
11. C A tropical low-pressure system with winds of 74 mph or more.
A.stratosphere B.conduction C.hurricane D.radiation
12. D Hottest layer of the atmosphere where the air is very thin, but can feel cold.
A.mesosphere B.nitrogen C.conduction D.thermosphere
13. C Energy that travels across distances as certain types of waves.
A.thermosphere B.oxygen C.radiation D.atom
14. B A large volume of air that has nearly the same temperature and humidity at different locations at the same altitude.
A.nitrogen B.air mass C.mesosphere D.ozone
15. D The transfer of energy from place to place by the motion of heated gas or liquid.
A.mesosphere B.atom C.stratosphere D.convection

Integrated Science Chapter 19 Review Answers

SA Adler



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oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$ reduction $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$ 24 3 $\text{Ag} + \text{Ni} \rightarrow \text{Ni}^{3+} + 3\text{Ag}$ oxidation reduction 25 $\text{Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$ oxidation reduction 26 O_2

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation state and oxidation number concepts to identify redox reactions and to keep track of electrons

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web jun 28 2010 answers i determine what is oxidized and what in each reaction identify the oxidizing agent and the reducing agent also $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$ Sr^0 to Sr^2 oxidized reducing agent O^0 to O^2 reduced ox ag $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$ Li^0 to Li^1 oxidized red ag S^0 to S^2 reduced ox ag 3 $2\text{Cs} + \text{Br}_2$