



Name: Steven Clarke

Date: 05-02-2022

Student Exploration: Collision Theory

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: activated complex, catalyst, chemical reaction, concentration, enzyme, half-life, molecule, product, reactant, surface area

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Suppose you added a spoonful of sugar to hot water and another to ice-cold water. Which type of water will cause the sugar to dissolve more quickly?

The hot water

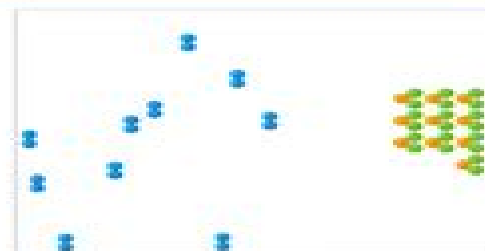
2. Suppose you held a lighted match to a solid hunk of wood and another match to a pile of wood shavings. Which form of wood will catch fire more easily?

Timber shavings

Gizmo Warm-up

A **chemical reaction** causes the chemical compositions of substances to change. **Reactants** are substances that enter into a reaction, and **products** are substances produced by the reaction. The *Collision Theory* Gizmo allows you to experiment with several factors that affect the rate at which reactants are transformed into products in a chemical reaction.

You will need blue, green, and orange markers or colored pencils for the first part of this activity.

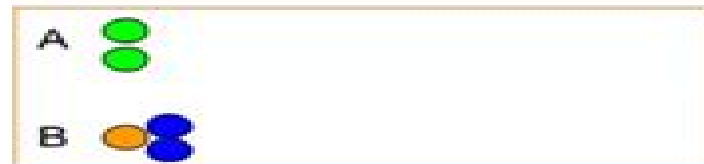


1. Look at the key at the bottom of the SIMULATION pane. In the space below, draw () the two reactants and two products of this chemical reaction.

Reactants:



Products:



2. Click **Play** (). What do you see?

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L Reisser

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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