

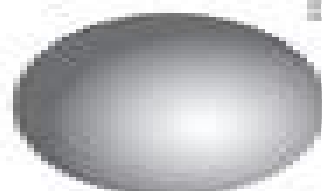
Name: _____

Class: _____

Date: _____

Identify which atomic model is shown in each picture and write that name below the picture. Write the scientist responsible for the model below each picture as well. There is a word bank below to help you.

History of the Atom

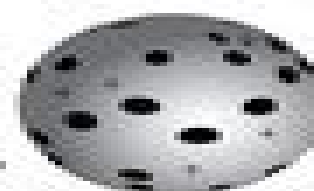


Indivisible atom

1a. Model: _____

b. Scientist(s) responsible: _____

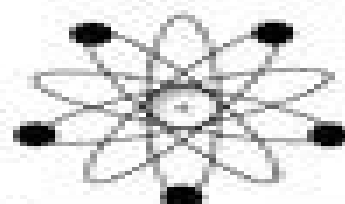
Scattered negatively charged small particles within the body



positively charged low density, spacious body

2a. Model: _____

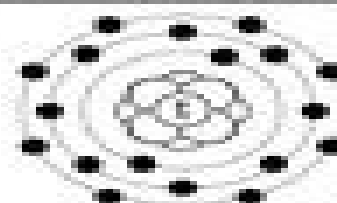
b. Scientist(s) responsible: _____



positively charged center with orbiting negatively charged particles

3a. Model: _____

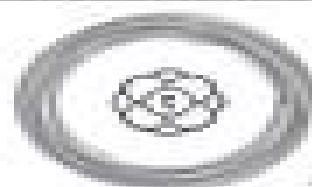
b. Scientist(s) responsible: _____



Nucleus with protons, neutrons, and electrons must orbit at defined energy levels and electrons away from the nucleus

4a. Model: _____

b. Scientist(s) responsible: _____



Nucleus with electrons that have calculated probable areas where they are located, but no exact location can be determined

5a. Model: _____

b. Scientist(s) responsible: _____

6. Now examine the following statement's from Dalton's Atomic Theory. Circle true or false for each statement and if it's false, write below what you know about the atom today that contradicts his theory

A. All matter is made of atoms.
True or False

B. Atoms are indivisible and indestructible.
True or False

C. All atoms of a given element are identical in mass and properties.
True or False

D. Compounds are formed by a combination of two or more different kinds of atoms.
True or False

Models that is chronological order 2
 primitive model
 simple, first atomic theory
 plum pudding model
 Bohr model
 Electron Cloud model

Scientists that is order 2
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 J.J. Thomson
 John Dalton
 Modern Scientists
 Niels Bohr

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Table of Contents History Of Atom Worksheet Answers

1. Understanding the eBook History Of Atom Worksheet Answers
 - The Rise of Digital Reading History Of Atom Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying History Of Atom Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an History Of Atom Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from History Of Atom Worksheet Answers
 - Personalized Recommendations
 - History Of Atom Worksheet Answers User Reviews and Ratings
 - History Of Atom Worksheet Answers and Bestseller Lists

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 - History Of Atom Worksheet Answers Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of History Of Atom Worksheet Answers
 - Highlighting and Note-Taking History Of Atom Worksheet Answers
 - Interactive Elements History Of Atom Worksheet Answers
8. Staying Engaged with History Of Atom Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers History Of Atom Worksheet Answers
9. Balancing eBooks and Physical Books History Of Atom Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection History Of Atom Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine History Of Atom Worksheet Answers
 - Setting Reading Goals History Of Atom Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of History Of Atom Worksheet Answers
 - Fact-Checking eBook Content of History Of Atom Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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