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Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

If Nate were to give Clara one of his markers, then both would end up with eight.

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

If Maggie gives Fred and Florence one marker each, they will all have eight markers.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *Ionic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** () to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain **electrons**. The outermost electrons are called **valence electrons**.

How many valence electrons does each atom have? Sodium: 1 Chlorine:

7

2. Click **Pause** (). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

Explore Learning Ionic Bonds Answers

Lauren Gardner



Explore Learning Ionic Bonds Answers:

Unveiling the Energy of Verbal Art: An Mental Sojourn through **Explore Learning Ionic Bonds Answers**

In a world inundated with screens and the cacophony of instant communication, the profound power and mental resonance of verbal artistry often diminish in to obscurity, eclipsed by the constant barrage of noise and distractions. However, situated within the musical pages of **Explore Learning Ionic Bonds Answers**, a captivating perform of fictional elegance that pulses with organic thoughts, lies an wonderful trip waiting to be embarked upon. Penned with a virtuoso wordsmith, that mesmerizing opus guides readers on a mental odyssey, lightly exposing the latent possible and profound affect stuck within the complicated internet of language. Within the heart-wrenching expanse with this evocative examination, we will embark upon an introspective exploration of the book is key subjects, dissect their interesting publishing model, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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