

CBSE Science - Grade X
Solution for 2022-23 Examination

(b) Why are some metal oxides categorized as amphoteric? Give One example.

(c) Complete the following equations :



OR

(c) On burning Sulphur in oxygen a colourless gas is produced.

(i) Write chemical equation for the reaction.

(ii) Name the gas formed.

(iii) State the nature of the gas.

(iv) What will be the action of this on a dry litmus paper?

Answer:

(a) Brown copper reacts with oxygen to form black copper(II) oxide.



(b) Some metal oxides such as aluminium oxide and zinc oxide, reacts with both acids and bases. So they are called amphoteric oxides.

(c) (i)



(ii)



OR

(c)



(ii) Sulphur dioxide

(iii) Sulphur dioxide is acidic in nature.

(iv) Sulphur dioxide turns moist blue litmus paper red. It does not have any action on dry litmus paper.

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