



# Manual Hp Elitebook 8440p

**D Kirk**



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web the addition of formaldehyde to amino acids and the resulting increase in their acidity towards phenol phthalein or thymol phthalein constitutes the basis of the well known sorenson formol method 2 for their estimation by titration with soda accordingly some knowledge of the acid base

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web the principle of this method is that formaldehyde make blocking of amino group of amino acid so amino acid become more acidic monomethylol and dimethylol derivative of amino acid and titrated

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web presence of a large excess formaldehyde amino acid can be titrated to a phenolphthalein end point when amino acid reacts with formaldehyde  $\text{NH}_2$  amino group in the amino acid will react with formaldehyde to form methylol and dimethylol derivatives which in

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web the amino acid solutions were at approximately pH 6 prior to the addition of formaldehyde the end point of the titration was at about pH 9 no correction was made for the volume of standard base required to bring the aqueous formaldehyde solution to the end point pH of the amino acid titration and the concentration of the formaldehyde

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