

# Calculating Half-Life Problems

Name \_\_\_\_\_ Block \_\_\_\_\_

## How to Calculate Half-Life Problems:

### THE PROBLEM:

An isotope of cesium (cesium-137) has a half-life of 30 years. If 1.0 g of cesium-137 disintegrates over a period of 90 years, how many grams of cesium-137 would remain?

### THE SOLUTION:

1. Draw a T-Table:

|  |  |
|--|--|
|  |  |
|  |  |
|  |  |
|  |  |

2. Label the left side with the unit of time mentioned in the problem.

| TIME (yrs) |  |
|------------|--|
|            |  |
|            |  |
|            |  |

3. Label the right side with the mass mentioned in the problem.

| TIME (yrs) | MASS (g) |
|------------|----------|
|            |          |
|            |          |
|            |          |

4. Begin by always writing a zero in the **TIME COLUMN**.

| TIME (yrs) | MASS (g) |
|------------|----------|
| 0          |          |
|            |          |
|            |          |

9. How many times you added a half-life in the **TIME COLUMN** equals how many half-lives have occurred.

10. The last amount of mass at the bottom of the **MASS COLUMN** equals how much mass is left after radioactive decay has occurred.

5. Then, in the **TIME COLUMN** add one half-life at a time till you reach the total time given in the problem.

| TIME (yrs) | MASS (g) |
|------------|----------|
| 0          |          |
| 30         |          |
| 60         |          |
| 90         |          |

6. In the **MASS COLUMN**, always start with the mass originally given in the problem.

| TIME (yrs) | MASS (g) |
|------------|----------|
| 0          | 1.0      |
| 30         |          |
| 60         |          |
| 90         |          |

7. Then keep dividing the number in the **MASS COLUMN** by 2 for each number of half-lives on the left column.

| TIME (yrs) | MASS (g) |
|------------|----------|
| 0          | 1.0      |
| 30         | .50      |
| 60         | .25      |
| 90         | .125     |

8. The rules are:

- Add half-lives on the left.
- Divide by 2 on the right.

|               |             |
|---------------|-------------|
|               |             |
| Add Half-Life | Divide by 2 |

# Instructional Fair Half Life Calculations Key

**David Baud**



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