

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Lab Section: \_\_\_\_\_

### Experiment 2: Molecular Models REPORT SHEET

## Pre-lab Study Questions:

- Objectives of the experiment (in your own words):
- What kinds of electrons form the covalent bonds in molecules?
- Is a polar molecule symmetrical or asymmetrical?

| FORMULA OF MOLECULE       | LEWIS STRUCTURE                                                                                                                               | 3-D SHAPE w/ lone pairs                                                                                                              | MOLECULAR GEOMETRY | MOLECULAR POLARITY |
|---------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------------------------------------------------------|--------------------|--------------------|
| example:<br>$\text{CH}_4$ | $\begin{array}{c} \text{H} \\   \\ \text{H}-\text{C}-\text{H} \\   \\ \text{H} \end{array}$                                                   |                                                                                                                                      | tetrahedral        | nonpolar           |
| $\text{H}_2\text{O}$      | $\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad \diagup \\ \text{O} \\ \diagup \quad \diagdown \\ \text{H} \end{array}$          | $\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad \diagup \\ \text{O} \\ \diagup \quad \diagdown \\ \text{H} \end{array}$ | tetrahedral        | polar              |
| $\text{H}_2\text{O}^+$    | $\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad \diagup \\ \text{O} \\ \diagup \quad \diagdown \\ \text{H} \end{array}$          | $\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad \diagup \\ \text{O} \\ \diagup \quad \diagdown \\ \text{H} \end{array}$ | tetrahedral        | nonpolar           |
| $\text{C}_2\text{H}_6$    |                                                                                                                                               |                                                                                                                                      |                    |                    |
| $\text{C}_2\text{H}_4$    | $\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ \text{H} \end{array}$ |                                                                                                                                      | trigonal planar    | non polar          |
| $\text{C}_2\text{H}_2$    |                                                                                                                                               |                                                                                                                                      |                    |                    |
| $\text{CO}_2$             | $\begin{array}{c} \text{O} \quad \text{O} \\ \diagdown \quad \diagup \\ \text{C} \\ \diagup \quad \diagdown \\ \text{O} \end{array}$          |                                                                                                                                      | Linear             | Non polar          |
| $\text{H}_2\text{O}_2$    |                                                                                                                                               |                                                                                                                                      |                    |                    |
| $\text{CHCl}_3$           |                                                                                                                                               |                                                                                                                                      |                    |                    |

# Molecular Model Laboratory Manual Answers

**Michael Rajnik**



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