# MATTER AND CHANGE

#### SECTION 2.1 PROPERTIES OF MATTER (pages 39-42)

This section helps you distinguish extensive from intensive properties and identify substances by their properties. It teaches you how to differentiate the three states of matter. It also defines a physical property and lists examples of pleysical properties and physical charges.

#### ▶ Describing Matter (page 39)

- of an object is a measure of the amount of matter the object contains.
- 2. How does an extensive property differ from an intensive property?

An extensive property depends on the amount of matter, while an intensive property depends on the type of matter.

without

#### ▶ Identifying Substances (page 40)

3. Matter that has a uniform and definite composition is called a

- 4. Is the following sentence true or false? All samples of a substance have Table 1 different physical properties.
- 5. A physical property is a quality or condition of a substance that can be measured

changing the substance's composition.

6. Circle the letter of the term that is NOT a physical property.

COLD ....

- a. bandos ss
- e. besilving posicyt.
- by contain
- d. m elting
- 7. Look at Table 2.1 on page 40. What is the melting point of bromine? \_\_\_\_\_\_\_
- 8. Look at Table 2.1 on page 40. Circle the letter of the substance that is a yellow solid and melts at 115°C.

on exhalthour

- b. chlorine
- c. gold:
- d. copper

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**Huangqi Zhang** 

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#### **Table of Contents Matter And Change Workbook Pearson**

- 1. Understanding the eBook Matter And Change Workbook Pearson
  - The Rise of Digital Reading Matter And Change Workbook Pearson
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Matter And Change Workbook Pearson
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Matter And Change Workbook Pearson
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Matter And Change Workbook Pearson
  - Personalized Recommendations
  - Matter And Change Workbook Pearson User Reviews and Ratings
  - Matter And Change Workbook Pearson and Bestseller Lists
- 5. Accessing Matter And Change Workbook Pearson Free and Paid eBooks

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- Matter And Change Workbook Pearson Budget-Friendly Options
- 6. Navigating Matter And Change Workbook Pearson eBook Formats
  - o ePub, PDF, MOBI, and More
  - Matter And Change Workbook Pearson Compatibility with Devices
  - Matter And Change Workbook Pearson Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Matter And Change Workbook Pearson
  - Highlighting and Note-Taking Matter And Change Workbook Pearson
  - Interactive Elements Matter And Change Workbook Pearson
- 8. Staying Engaged with Matter And Change Workbook Pearson
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Matter And Change Workbook Pearson
- 9. Balancing eBooks and Physical Books Matter And Change Workbook Pearson
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Matter And Change Workbook Pearson
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Matter And Change Workbook Pearson
  - Setting Reading Goals Matter And Change Workbook Pearson
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Matter And Change Workbook Pearson
  - Fact-Checking eBook Content of Matter And Change Workbook Pearson
  - Distinguishing Credible Sources
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  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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