

CHAPTER 13: KINETICS: LEARNING GOALS

Learning Goal 1: Reaction Rates

Pages 13.1-13.2

Reaction Rate: The change in concentration of a reactant or product over time.

Initial Rate: The rate of reaction at the beginning of the reaction. It is the slope of the tangent line to the concentration vs. time curve at $t = 0$.

Key Terms:

- Reaction rate: the change in concentration of a reactant or product over time.
- Initial rate: the rate of reaction at the beginning of the reaction.
- Rate of reaction: the change in concentration of a reactant or product over time.
- Rate of reaction: the change in concentration of a reactant or product over time.
- Rate of reaction: the change in concentration of a reactant or product over time.

Learning Goal 2: Reaction Mechanisms

Reaction Mechanism: A series of steps that describe the overall reaction.

Key Terms:

Reaction mechanism: A series of steps that describe the overall reaction. The rate-determining step is the slowest step in the reaction. The overall reaction is the sum of the steps in the reaction.

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DJ Losen



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