

## Section I: Multiple-Choice Questions

Time: 90 minutes

100 questions

**Directions:** Each of the questions or incomplete statements in this section is followed by five answer choices. Select the best answer choice and fill in the corresponding circle on the answer sheet.

- Which of the following is responsible for reducing global CFC production by half?
  - Copenhagen Protocol
  - Montreal Protocol
  - Pollution Prevention Act
  - Kyoto Protocol
  - Convention on Long-Range Transboundary Air Pollution
- The main global contributing factors to greenhouse-gas emissions are
  - Electricity production and transportation
  - Industrial processes
  - Agricultural processes
  - Agriculture and transportation
  - None of the above
- Which of the following is a way to potentially avoid a tragedy of the commons?
  - People sharing the resource work to manage the resource
  - Private ownership
  - Regulation by the government
  - All of the above
  - None of the above
- In the transitional stage of demographic transition, why does the death rate fall?
  - Movement of people into urban areas
  - Less disease on a global scale
  - More efficient and reliable food production and medical advancements
  - Fewer births
  - Fewer work-related accidents
- What is the most abundant greenhouse gas in the atmosphere?
  - Methane
  - Nitrogen dioxide
  - Water vapor
  - Carbon dioxide
  - Sulfur dioxide
- How does a nuclear meltdown occur?
  - Radiation is released when temperatures in the core of the reactor increase and the uranium fuel rods melt.
  - Radiation escapes from the core of the reactor due to the shutdown of the generator.
  - The nuclear core collapses.
  - The fission process halts instantaneously.
  - None of the above.
- As a whole, industrialized nations use \_\_\_\_\_% more energy than developing nations.
  - 10
  - 25
  - 50
  - 75
  - 100
- $$\text{N}_2 + \text{O}_2 \rightarrow 2\text{NO}$$

$$2\text{NO} + \text{O}_2 \rightarrow 2\text{NO}_2$$

$$\text{NO}_2 \rightarrow \text{NO} + \text{O}$$

$$\text{O} + \text{O}_2 \rightarrow \text{O}_3$$

The above reactions describe the chemical process that forms:

  - Photochemical smog
  - Industrial smog
  - Acid precipitation
  - Ozone
  - Peroxyacetyl nitrate

# Environmental Science Practice Test 1 Answers Key

**The Princeton Review**



## **Environmental Science Practice Test 1 Answers Key:**

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