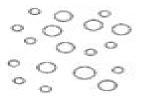
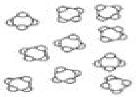
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Answer the following as completely as possible. Use your notebook assignments to help you prepare for your second test. If you cannot find the information in your notebook, than refer to your textbook.

- Describe a heterogeneous mixture and give an example. A mixture where you can see different parts example M&Ms
- Describe a homogeneous mixture and give an example. A mixture where the pieces appear to look similar throughout because substances are so evenly mixed. An example is Salt water
- 3. What makes up elements? A pure substance that cannot be broken down into any other substance
- 4. What makes up compounds? a substance made of two or more elements
- What is a mixture? Made up of more than one kind of matter Can be separated physically by: Evaporation, Filtering, Sorting, Electric Charge
- Label the pictures as compound, mixture or element. Each circle represents an atom. If two atoms are touching then they are bonded together. #1 Element, #2 mixture, #3 compound







- 7. What is a Physical Property? Properties that do not change the chemical nature of matter
- What is a Chemical Property? Properties that describe the way a substance may change or react to form another substance
- 9. Give Examples of:

Physical Properties	Chemical Properties
Color-Odor-Luster- Malleability- Conductivity- Hardness- Melting Point/Freezing Point- Boiling Point- Density- Volume-Length-	Combustibility-reactivity

- Give examples of phase changes. Melting, boiling, evaporation, condensation,
- 11. What is Matter? Matter is anything that has mass and volume, It occupies space, composed of atoms
- 12. How can matter be broken down into smaller groups? A chart is fine for this one.

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Table of Contents Mixtures And Substances Study Guide Key

- 1. Understanding the eBook Mixtures And Substances Study Guide Key
 - The Rise of Digital Reading Mixtures And Substances Study Guide Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Mixtures And Substances Study Guide Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mixtures And Substances Study Guide Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Mixtures And Substances Study Guide Key
 - Personalized Recommendations
 - Mixtures And Substances Study Guide Key User Reviews and Ratings
 - Mixtures And Substances Study Guide Key and Bestseller Lists
- 5. Accessing Mixtures And Substances Study Guide Key Free and Paid eBooks
 - Mixtures And Substances Study Guide Key Public Domain eBooks
 - Mixtures And Substances Study Guide Key eBook Subscription Services
 - Mixtures And Substances Study Guide Key Budget-Friendly Options
- 6. Navigating Mixtures And Substances Study Guide Key eBook Formats
 - o ePub, PDF, MOBI, and More
 - Mixtures And Substances Study Guide Key Compatibility with Devices
 - Mixtures And Substances Study Guide Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mixtures And Substances Study Guide Key
 - Highlighting and Note-Taking Mixtures And Substances Study Guide Key
 - Interactive Elements Mixtures And Substances Study Guide Key
- 8. Staying Engaged with Mixtures And Substances Study Guide Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Mixtures And Substances Study Guide Key
- 9. Balancing eBooks and Physical Books Mixtures And Substances Study Guide Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mixtures And Substances Study Guide Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mixtures And Substances Study Guide Key
 - Setting Reading Goals Mixtures And Substances Study Guide Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mixtures And Substances Study Guide Key
 - Fact-Checking eBook Content of Mixtures And Substances Study Guide Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - o Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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