



I Pod 3 User Guide

Robert R. Redfield



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web cause the moose and wolf populations to fluctuate over time the isle royale model in ecobeaker in this lab you will explore populations of predators and prey using a simplified simulation model of the isle royale system the isle royale model involves three species plants moose and wolves

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web wolf moose populations 1980 to today the populations of wolves and moose on isle royale constantly adjust as food availability weather and seasons affect the health and reproduction of the animals over time notice as the wolf population declines the moose population increases and vice versa in the past forty years isle royale has

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web to analyze the relationship of predator prey interactions between moose and wolves what is population growth changes in the size and composition of populations and the factors that cause those changes fill in the blanks the larger a population becomes the it

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involved in this activity wolf and moose wolf and moose 2 what is the purpose of this study for example how long has this study been happening why did it start what are the goals 3

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web provide a biological explanation for why the moose population overshoots its carrying capacity when moose first colonize isle royale hint consulting the introduction might help at year 50 or later with the moose population at its carrying capacity what would happen if an extra 200 moose suddenly arrived on isle royale how would this

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web where the salt is the conjugate base of the acid the four parts of the titration curve are described below and you should look to the appropriate text section to see how they are treated pure acid 0 ml of base is added section 17.3.2.1

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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of hcl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0 100 m the ph of the solution is then $\text{ph} = -\log 0.100 = 1.000$ ph log 0 100 1 000 ph log 0 100 1 000 b titrant volume 12 50 ml

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web expert answer 100 2 ratings the example is based upon henderson hasselbalch equation the equation gives relation between ph and pha of weak acid the equation is $\text{ph} = \text{pka} + \log \frac{\text{salt}}{\text{acid}}$ at equivalence point all a view the full answer transcribed image text

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web ab acid base titration curves purpose the purpose of this lab is to determine the shape of acid base titration curves theoretically and experimentally and to use an acid base titration to measure the effectiveness of antacid tablets summary a titration curve is the plot of the ph value of a solution vs the amount of titrant added

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web strong acids and strong bases completely ionize in solution resulting in water and a salt for a weak acid there is only partial ionization the titration curve of a strong base strong acid showed a stretched out curve as it started with a slow gradual change in ph as it reached the equivalence point

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