

Chapter 6 Practice Test: Chemical Bonding

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The charge on an ion is
a. always positive.
b. always negative.
c. either positive or negative.
d. zero.
- _____ 2. According to the octet rule, a calcium atom has a tendency to
a. lose one electron.
b. lose two electrons.
c. gain one electron.
d. gain two electrons.
- _____ 3. If a compound forms by ionic bonding, which is *not* true?
a. A positively charged atom or group of atoms attracts a negatively charged atom or group of atoms.
b. The net charge of the compound is zero.
c. The compound contains just two atoms, each of opposite charge.
d. Several ions group together in a tightly packed structure.
- _____ 4. The only property listed that is *not* characteristic of ionic compounds is
a. high melting point.
b. hardness.
c. lack of crystal structure.
d. good conductor of electricity.
- _____ 5. Which formula listed below represents a polyatomic ion?
a. NH_4^+
b. H_2SO_4
c. NaCl
d. H_2O
- _____ 6. The melting points of ionic compounds are higher than the melting points of molecular compounds because
a. ionic substances tend to vaporize at room temperature.
b. ionic substances are brittle.
c. attractive forces between ions are greater than the attractive forces between molecules.
d. the numbers of positive and negative charges are equal in an ionic compound.

Modern Chemistry Chapter 4 Review

A.E. Derome



Modern Chemistry Chapter 4 Review:

Decoding **Modern Chemistry Chapter 4 Review**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Modern Chemistry Chapter 4 Review**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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Table of Contents Modern Chemistry Chapter 4 Review

1. Understanding the eBook Modern Chemistry Chapter 4 Review
 - The Rise of Digital Reading Modern Chemistry Chapter 4 Review
 - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Chapter 4 Review
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Chapter 4 Review
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Chapter 4 Review
 - Personalized Recommendations
 - Modern Chemistry Chapter 4 Review User Reviews and Ratings
 - Modern Chemistry Chapter 4 Review and Bestseller Lists

5. Accessing Modern Chemistry Chapter 4 Review Free and Paid eBooks
 - Modern Chemistry Chapter 4 Review Public Domain eBooks
 - Modern Chemistry Chapter 4 Review eBook Subscription Services
 - Modern Chemistry Chapter 4 Review Budget-Friendly Options
6. Navigating Modern Chemistry Chapter 4 Review eBook Formats
 - ePub, PDF, MOBI, and More
 - Modern Chemistry Chapter 4 Review Compatibility with Devices
 - Modern Chemistry Chapter 4 Review Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modern Chemistry Chapter 4 Review
 - Highlighting and Note-Taking Modern Chemistry Chapter 4 Review
 - Interactive Elements Modern Chemistry Chapter 4 Review
8. Staying Engaged with Modern Chemistry Chapter 4 Review
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Chemistry Chapter 4 Review
9. Balancing eBooks and Physical Books Modern Chemistry Chapter 4 Review
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modern Chemistry Chapter 4 Review
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modern Chemistry Chapter 4 Review
 - Setting Reading Goals Modern Chemistry Chapter 4 Review
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modern Chemistry Chapter 4 Review
 - Fact-Checking eBook Content of Modern Chemistry Chapter 4 Review
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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