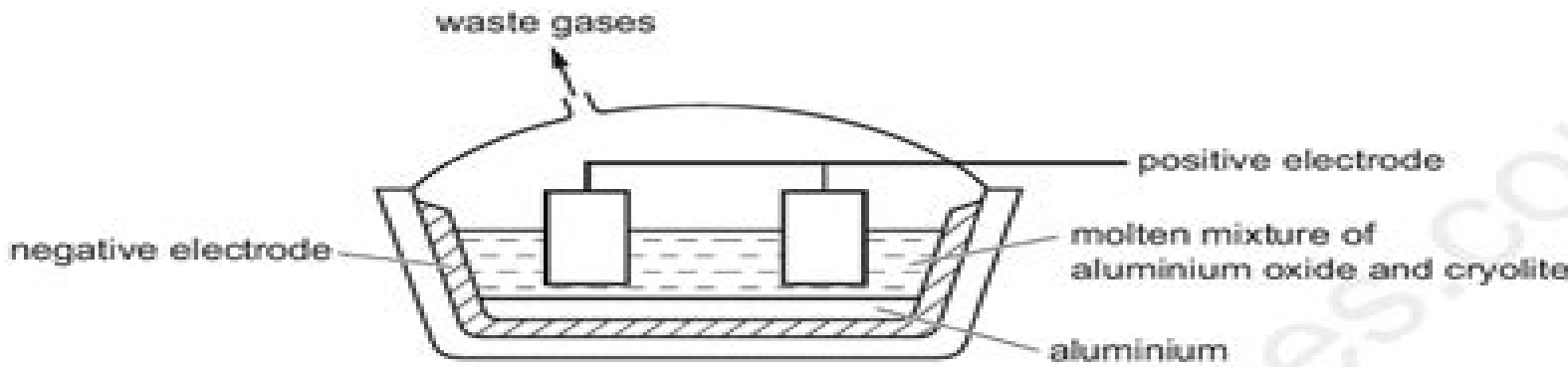


3 (a) Name the ore of aluminium which mainly consists of aluminium oxide.

Bauxite [1]

(b) Aluminium is produced by the electrolysis of aluminium oxide dissolved in molten cryolite.



(i) Give **two** reasons why the electrolysis is done using a molten mixture of aluminium oxide and cryolite instead of molten aluminium oxide only.

- 1 It improves conductivity or It is a better conductor
 - 2 It has a lower operating temperature.
- [2]

(ii) Write ionic half-equations for the reactions occurring at the electrodes.

positive electrode $2\text{O}^{2-} \rightarrow \text{O}_2 + 4\text{e}^-$

negative electrode $\text{Al}^{3+} + 3\text{e}^- \rightarrow \text{Al}$

[2]

(iii) The anodes are made of carbon and have to be replaced regularly.

Explain why the carbon anodes have to be replaced regularly.

Anodes(or carbon) react with oxygen to form carbondioxide.Hence they need to be replaced.

[2]

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Table of Contents Igcse Past Papers Chemistry Multiple Choice

1. Understanding the eBook Igcse Past Papers Chemistry Multiple Choice
 - The Rise of Digital Reading Igcse Past Papers Chemistry Multiple Choice
 - Advantages of eBooks Over Traditional Books
2. Identifying Igcse Past Papers Chemistry Multiple Choice
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Igcse Past Papers Chemistry Multiple Choice
 - User-Friendly Interface
4. Exploring eBook Recommendations from Igcse Past Papers Chemistry Multiple Choice
 - Personalized Recommendations
 - Igcse Past Papers Chemistry Multiple Choice User Reviews and Ratings
 - Igcse Past Papers Chemistry Multiple Choice and Bestseller Lists

5. Accessing Igcse Past Papers Chemistry Multiple Choice Free and Paid eBooks
 - Igcse Past Papers Chemistry Multiple Choice Public Domain eBooks
 - Igcse Past Papers Chemistry Multiple Choice eBook Subscription Services
 - Igcse Past Papers Chemistry Multiple Choice Budget-Friendly Options
6. Navigating Igcse Past Papers Chemistry Multiple Choice eBook Formats
 - ePub, PDF, MOBI, and More
 - Igcse Past Papers Chemistry Multiple Choice Compatibility with Devices
 - Igcse Past Papers Chemistry Multiple Choice Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Igcse Past Papers Chemistry Multiple Choice
 - Highlighting and Note-Taking Igcse Past Papers Chemistry Multiple Choice
 - Interactive Elements Igcse Past Papers Chemistry Multiple Choice
8. Staying Engaged with Igcse Past Papers Chemistry Multiple Choice
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Igcse Past Papers Chemistry Multiple Choice
9. Balancing eBooks and Physical Books Igcse Past Papers Chemistry Multiple Choice
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Igcse Past Papers Chemistry Multiple Choice
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Igcse Past Papers Chemistry Multiple Choice
 - Setting Reading Goals Igcse Past Papers Chemistry Multiple Choice
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Igcse Past Papers Chemistry Multiple Choice
 - Fact-Checking eBook Content of Igcse Past Papers Chemistry Multiple Choice
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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