

### **Ionic and Covalent Naming Practice **Answers****

*For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.*

- |     |                           |                                  |
|-----|---------------------------|----------------------------------|
| 1)  | $\text{Na}_2\text{CO}_3$  | <b>sodium carbonate</b>          |
| 2)  | $\text{P}_4\text{O}_{10}$ | <b>tetraphosphorus decaoxide</b> |
| 3)  | $\text{N}_2\text{O}_5$    | <b>dinitrogen pentoxide</b>      |
| 4)  | $\text{FeSO}_4$           | <b>iron (II) sulfate</b>         |
| 5)  | $\text{SiO}_2$            | <b>silicon dioxide</b>           |
| 6)  | $\text{FeCl}_3$           | <b>iron (III) chloride</b>       |
| 7)  | $\text{CoBr}_2$           | <b>cobalt (II) bromide</b>       |
| 8)  | $\text{B}_2\text{H}_4$    | <b>diboron tetrahydride</b>      |
| 9)  | $\text{CO}$               | <b>carbon monoxide</b>           |
| 10) | $\text{PCl}_3$            | <b>phosphorus trichloride</b>    |

*For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.*

- |     |                        |  |
|-----|------------------------|--|
| 11) | dinitrogen trioxide    | <b><math>\text{N}_2\text{O}_3</math></b>             |
| 12) | nitrogen dioxide       | <b><math>\text{NO}_2</math></b>                      |
| 13) | sulfur pentoxide       | <b><math>\text{SO}_5</math></b>                      |
| 14) | lithium acetate        | <b><math>\text{LiC}_2\text{H}_3\text{O}_2</math></b> |
| 15) | phosphorus trifluoride | <b><math>\text{PF}_3</math></b>                      |
| 16) | vanadium (V) oxide     | <b><math>\text{V}_2\text{O}_5</math></b>             |
| 17) | aluminum hydroxide     | <b><math>\text{Al}(\text{OH})_3</math></b>           |
| 18) | zinc sulfide           | <b><math>\text{ZnS}</math></b>                       |
| 19) | silicon tetrafluoride  | <b><math>\text{SiF}_4</math></b>                     |
| 20) | silver phosphate       | <b><math>\text{Ag}_3\text{PO}_4</math></b>           |

# Guided Answers Ionic Bonds Ionic Compounds

**JR Anderson**



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