# Unit 7: Chemical Formulas and Compounds Guided Notes

L C	h. 7.1	: Naming	lonic	Compound	S
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1.	Ch.	7.1: Na	ming	lonic C	ompounds		
	A. Warm Up:						
		1.	Distinguish between a cation and an anion.				
		2.	Which elements tend to form cations? anions?				
		3.	Using a periodic table, predict the charge ion each of the following elements form:				
			a)	colciu	on c) sulfur		
			ь)	fluorii	ne d) aluminum		
	В.	Writ	ing Forr	nulas an	nd Naming Chemical Compounds		
		1.	There	e are two	types of chemical compounds:		
			a)	ionic (	compounds		
			ь)	covale	ent (malecular) compounds		
	C.	Ionic Compounds are composed of a cation and an anion.					
		1	catio	ns – posi	tively charged ion ()		
		anions – negatively charged ion ()					
		3.	. Monoatomic lons				
			a) Cotions				
				(1)	For monatomic cations, the name of the ion is the name of the element.		
				(2)	examples:		
					(a) Na' =		
					(b) Mg <sup>2+</sup> =		
					(c) Al <sup>2+</sup> =		
			b)	Some	metals are able to form more than one stable ion ()		
				(1)	usually transition elements		
				(2)	example:		
				(3)	forms Cu <sup>2+</sup> and Cu <sup>+</sup>		
				(4)	use Roman numerals to distinguish charge		

(5) copper (II) and copper (I)

# **Guided And Study Chemical Names Formulas**

**RJ Alexander** 

**Guided And Study Chemical Names Formulas:** 

The Enigmatic Realm of Guided And Study Chemical Names Formulas: Unleashing the Language is Inner Magic

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