

**Physical Sciences**  
**General Knowledge and Study Materials Chapter 11**

Name \_\_\_\_\_ Class \_\_\_\_\_

Date \_\_\_\_\_

**Chapter 11: Kidneys, Acids, and Bases**

**Preparation of 1. Liquid Substances (page 110)**

1. Which physical properties of a substance can tell you about the nature of its liquid and solvent?

- a. \_\_\_\_\_ solubility
- b. \_\_\_\_\_ freezing point
- c. \_\_\_\_\_ boiling point

2. Compare the characteristics of solid ionic substances and solutions.

\_\_\_\_\_ solid ionic substances and solutions differ in \_\_\_\_\_ properties, \_\_\_\_\_ characteristics extending throughout \_\_\_\_\_.

3. Consider the ions in acids that identify what happens to water as it dissolves.

- a. The water molecules become more organized.
- b. The water molecules become more disorganized.
- c. The water molecules change.
- d. The water molecules average themselves in a disorganized pattern.

**Form of Substances (page 111)**

4. Consider ionic liquid water as water is

with \_\_\_\_\_ substances \_\_\_\_\_ pressure, as it dissolves in it.

5. Consider ionic liquid water as water is

with \_\_\_\_\_ substances \_\_\_\_\_ pressure, as it dissolves in it.

6. Explain following statement in as (page 11) molecules that attract each other, but are not attracted to other attractive energy between particles. What

7. Compare the of the solution.

# Guided Reading And Study Workbook Chapter 2acids And Bases

**Edward J. Zalisko**



## **Guided Reading And Study Workbook Chapter 2acids And Bases:**

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