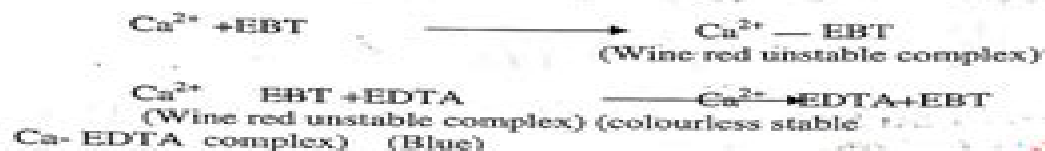
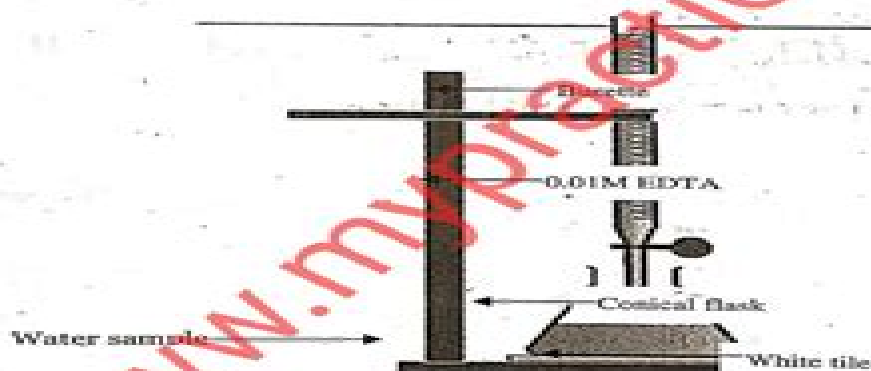


alkaline earth metal ions Ca^{2+} and Mg^{2+} . The EDTA reagent can be used to measure the total quantity of dissolved Ca^{2+} and Mg^{2+} ions in a water sample. Determination of hardness of water is based on the fact that when indicator, Eriochrome Black -T (EBT) is added to hard water in alkaline medium, it forms wine red colour complex with Ca^{2+} and Mg^{2+} ions.



Thus the total hardness of a water sample can be estimated by titration with standard solution of EDTA.

VIII Practical set-up / Circuit diagram / Work Situation



IX Resources Required

S. No.	Name of Resource	Suggested Broad Specification	Quantity
1.	Burette	Borosil glass , Capacity 25 ml /50 ml	One per group
2.	Conical flask	Borosil glass, Capacity 250 ml/100 ml	One per group
3.	Beaker	Borosil glass, Capacity 250 ml/100 ml	One per group
4.	Pipette	Borosil glass, Capacity 25ml / 10 ml,	One per group
5.	EDTA solution	0.01 M	One per group
6.	Buffer solution	pH = 10	As per Requirement
7.	Eriochrome Black T indicator		
8.	Water sample	Hard water sample	

Msbte Applied Chemistry Answer Key

G Orfield



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