Name Class

Assessment)

# Chapter Test A

# Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- Which of the properties listed below is not characteristic of an acid?
  - a. a sour taste
  - b. a slippery feel
  - c. the ability to conduct an electric current
  - d. reactivity with metals
  - 2. Which is an example of a weak acid?
    - a. HCI
    - **b.** H<sub>2</sub>SO<sub>4</sub>
    - c. HaPO<sub>4</sub>
    - d. HNO<sub>a</sub>
- \_\_\_\_\_ 3. Acetic acid is found in
  - a. vinegar.
  - b. the stomach.
  - c. antacids.
  - d. oranges.
- \_\_\_\_\_ 4. A characteristic of an Arrhenius base is that it
  - a. is an electrolyte.
  - b. tastes bitter.
  - c. dissociates to form OH ons.
  - d. All of the above
- 5. Which of the following is a Brønsted-Lowry base?
  - a. HCl
  - b. HCO<sub>a</sub>
  - c. H<sub>0</sub>O \*
  - d.  $H_0PO_4$

6. In the reaction represented by the equation

$$H_2C_2O_4(aq) + CH_3NH_2(aq) \Leftrightarrow HC_2O_4^-(aq) + CH_3NH_2^+(aq),$$

which of these is a conjugate acid-base pair?

- a.  $H_2C_2O_4(aq)$  and  $CH_3NH_2(aq)$
- b. H<sub>2</sub>C<sub>2</sub>O<sub>4</sub>(aq) and CH<sub>3</sub>NH<sup>\*</sup><sub>3</sub>(aq)
- c. CH<sub>3</sub>NH<sub>2</sub>(aq) and CH<sub>3</sub>NH'<sub>3</sub>(aq)
- d. HC2O4(aq) and CH2NH2(aq)

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