

Mole Conversions Worksheet

1. How many moles of magnesium is 3.01×10^{22} atoms of magnesium?

0.05 moles

2. How many molecules are there in 4.00 moles of glucose, $C_6H_{12}O_6$?

2.408×10^{24}

3. How many moles are 1.20×10^{25} atoms of phosphorous?

19.93 moles

4. How many atoms are in 0.750 moles of zinc?

4.515×10^{23}

5. How many molecules are in 0.400 moles of dinitrogen pentoxide?

2.408×10^{23}

11. Find the mass in grams of 2.00×10^{23} molecules of $C_{15}H_{13}O$ (called penguinone because its structure looks like a penguin).

50.17 g

12. Determine the mass, in grams, of 2.6 moles of angelic acid, $C_5H_8O_2$.

260 g

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