

MARONDERA UNIVERSITY

OF AGRICULTURAL SCIENCES AND TECHNOLOGY

ADMINISTRATION OFFICE

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Marondera Nursing Intake

J Ma



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 $1\text{ N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
 $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
 $4\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$
 $\text{Pb} + 2\text{HCl} \rightarrow \text{PbCl}_2 + \text{H}_2$
 $2\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3$
 $3\text{K}_2\text{SO}_4 + 6\text{KBr} + \text{Al}_2\text{SO}_4 \rightarrow 3\text{K}_2\text{SO}_4 + 3\text{H}_2\text{O} + 2\text{Al(OH)}_3$

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 $2\text{Fe}_2\text{O}_3 + 3\text{C} \rightarrow 4\text{Fe} + 3\text{CO}_2$
 on towards the next step now step 3 once you are done with this check out if the number of atoms for the other elements is also equal on both the sides

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web 1 78 348 stoichiometry helps us determine how much substance is needed or is present things that can be measured are reactants and products mass molecular weight chemical equations formulas stoichiometric coefficient the stoichiometric coefficient or stoichiometric number is the number of molecules that participate in the reaction

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web expert answer transcribed image text balancing equations and simple stoichiometry balance the following equations
 $\text{N}_2 + \text{F}_2 \rightarrow 2\text{NF}_3$
 $2\text{C}_6\text{H}_{10} + \text{O}_2 \rightarrow 2\text{CO}_2 + \text{H}_2\text{O}$
 $3\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $\text{GaBr}_3 + \text{NaBr} \rightarrow \text{SnO}_4 + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 5\text{NF}_3 + 2\text{SnF}_2 + \text{N}_2\text{O}_3$
 using the equation from problem 2 above answer the following

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 $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$
 $2\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$
 $3\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $2\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 6\text{NaBr} + 3\text{SnO}_2 + 3\text{NF}_3 + 3\text{SnF}_2 + \text{N}_2\text{O}_3$
 using the

following equation $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 $\text{Br}_2 + \text{H}_2\text{O} \rightarrow \text{HBr} + \text{HCO}_3^-$
 $\text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \rightarrow \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + \text{NaBr}$
 $\text{SNO}_2 + \text{NF}_3 \rightarrow \text{SNO}_3 + \text{NF}_2$
 using the following equation
 $\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4$

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 $\text{N}_2 + \text{F}_2 + \text{NF}_3 + \text{C}_6\text{H}_{10} + \text{O}_2 + \text{CO} + \text{H}_2\text{O} + \text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2 + \text{GaBr} + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + \text{NaBr} + \text{SNO} + \text{NF}_3 + \text{SNO}_2$

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web may 28 2020 write the balanced equation for the reaction and calculate how much SiO_2 is required to produce 3.00 kg of SiC
 $\text{SiO}_2 + 3\text{C} \rightarrow \text{SiC} + 2\text{CO}$

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 $\text{N}_2 + \text{F}_2 + \text{NF}_3 + \text{C}_6\text{H}_{10} + \text{O}_2 + \text{CO} + \text{H}_2\text{O} + \text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2 + \text{GaBr} + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + \text{NaBr} + \text{SNO} + \text{NF}_3 + \text{SNO}_2$
 solve the following stoichiometry grams problems 6

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web gabrielle m 9 years ago i m working on chemical reactions double and single replacement on flvs now my first question for this video is how do you have 4 aluminum atoms when it says 2Al subscript 2 do you just add the 2 and the subscript 2 same thing goes with the o3 i am so lost

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web google classroom balance the following chemical equation $\text{Mg} + \text{OH}^- \rightarrow \text{Mg}^{2+} + \text{OH}^-$ $\text{MgCl}_2 + \text{H}_2\text{O} \rightarrow \text{Mg}^{2+} + \text{OH}^-$ note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem $7.4 \times 10^5 \theta \pi 8.5 \times 10^6$

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mar 16 2023 ingredients 500g strong white bread flour 8g fine salt 10g fast action dried yeast 30ml olive oil plus extra for oiling 370ml water 75g pitted kalamata black olives 1 small red onion 10 cherry tomatoes 1 tbsp dried oregano about 150ml olive oil to drizzle flaky sea salt method 1

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put the loaf tray on the middle shelf and bake for 25 minutes then lower the oven to 200 c 392 f gas mark 6 and bake for 10 15 minutes more until the crust has a good colour hold the loaf in a tea towel and tap the bottom if it sounds hollow it s

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ingredients 500g 1lb 2oz strong flour 15g ½oz salt 55ml 2fl oz olive oil 20g ¾oz fresh yeast 275ml 9fl oz water how to videos method mix all the ingredients in a large bowl taking care not to put

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meanwhile heat the oven to 230 c 210 c fan 450 f gas 8 and put a roasting tray in the bottom of the oven to heat up step 9 remove the risen loaf from the bag and dust with flour fill the roasting tray with cold water to create steam and put the bread in the oven bake for 15 minutes then lower the oven to 190 c 170 c fan 375 f gas 5

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