

# Balancing Chemical Equations (Medium)

## Answer Sheet

### Question Sheet

Balancing Chemical Equations 2

**Chemistry**

Name: \_\_\_\_\_

Balance the following chemical equations.

1.  $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$

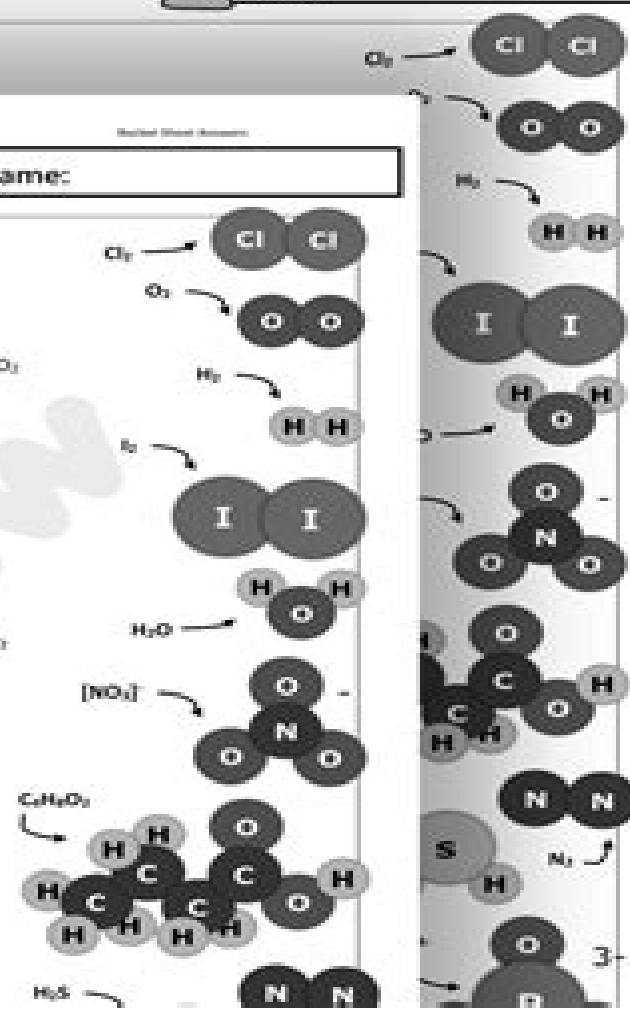
Balancing Chemical Equations 2

**Chemistry**

Name: \_\_\_\_\_

Balance the following chemical equations.

- $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
- $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- $\text{CaCO}_3 + \text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$
- $\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$
- $\text{Ca} + \text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$
- $\text{NO} + \text{H}_2\text{O} + \text{O}_2 \rightarrow \text{HNO}_3$
- $\text{Al}(\text{OH})_3 + \text{HNO}_3 \rightarrow \text{H}_2\text{O} + \text{Al}(\text{NO}_3)_3$
- $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- $\text{HCl} + \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$
- $\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2\text{O}$
- $\text{Ca}(\text{OH})_2 + \text{HNO}_3 \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2\text{O}$
- $\text{Ba}(\text{HCO}_3)_2 \rightarrow \text{BaCO}_3 + \text{H}_2\text{O} + \text{CO}_2$
- $\text{MgBr}_2 + \text{KOH} \rightarrow \text{KBr} + \text{Mg}(\text{OH})_2$
- $\text{K} + \text{H}_2\text{O} \rightarrow \text{KOH} + \text{H}_2$
- $\text{C}_2\text{H}_5\text{O}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{CO}_2$
- $\text{Rb} + \text{RbNO}_3 \rightarrow \text{Rb}_2\text{O} + \text{N}_2$



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