

Assessment

Quiz

Section: Describing Chemical Reactions

In the space provided, write the letter of the term or phrase that best answers the question.

- _____ 1. The formation of a gas is evidence of
a. a chemical change.
b. a physical change.
c. no change in energy.
d. Both (a) and (c).
- _____ 2. Which of the following may be evidence of a chemical change?
a. release of energy as heat
b. production of sound
c. reduction in temperature
d. All of the above
- _____ 3. In the word equation,
iron + copper(II) sulfate → iron(II) sulfate + copper,
a product is
a. iron(II) sulfate,
b. copper,
c. iron,
d. Both (a) and (b)
- _____ 4. In the reaction described by the following equation
water + energy → hydrogen gas + oxygen gas,
energy
a. can be considered a reactant.
b. is absorbed.
c. is released.
d. Both (a) and (b)
- _____ 5. In the unbalanced formula equation
carbon + hydrogen → methane, the formula
CH₄ represents
a. one atom of methane.
b. five atoms of methane.
c. one molecule of methane.
d. one gram of methane.

Holt Chemistry Section One Assessment Quiz Answers

L Darling-Hammond



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