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Chemistry Comes Alive

Objectives

Part 1: Basic Chemistry

Definition of Concepts: Matter and Energy

- Differentiate between matter and energy and between potential energy and kinetic energy.
- 2. Describe the major energy forms.

Composition of Matter: Atoms and Elements

- 3. Define chemical element and list the four elements that form the bulk of body matter.
- Define atom. List the subotomic particles, and describe their relative masses, charges, and positions in the atom.
- 5. Define atomic number, atomic mass, atomic weight, isotope, and radioisotope.

How Matter Is Combined: Molecules and Mixtures

- 6. Define molecule, and distinguish between a compound and a mosture.
- 7. Compare solutions, colloids, and suspensions.

Chemical Bonds

- 8. Explain the role of electrons in chemical bonding and in relation to the octet rule.
- 9. Differentiate among ionic, covulent, and hydrogen bonds.
- 10. Compare and contrast polar and nonpolar compounds.

Chemical Reactions

- 11. Define the three major types of chemical reactions: synthesis, decomposition, and exchange. Comment on the nature of oxidation-reduction reactions and their importance.
- 12. Explain why chemical reactions in the body are often irreversible

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